

LDA + U calculations using pwscf

LDA + U:

- What is it ?
- When is it useful ?
- How to use it ?

→ Examples on NiO and FeO

LDA + U

- LDA + U energy functional:

$$\begin{aligned} E^{\text{LDA}+U}[\rho^\sigma, n_\lambda^{I,\sigma}] &= E^{\text{LDA}}[\rho^\sigma] + E_U^{\text{Hub}}[n_\lambda^{I,\sigma}] - E_U^{\text{dc}}[n^{I,\sigma}] \\ &= E^{\text{LDA}}[\rho^\sigma] + \frac{U}{2} \sum_{I,\lambda,\sigma} n_\lambda^{I,\sigma} (1 - n_\lambda^{I,\sigma}) \end{aligned}$$

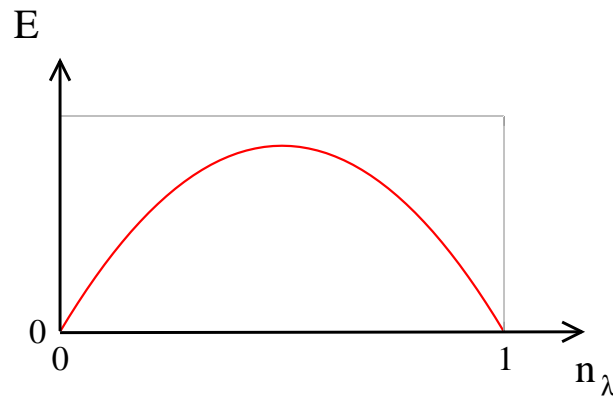
- $n_\lambda^{I,\sigma}$: occupation numbers of the localized (d or f) orbitals, obtained by diagonalizing atomic occupation matrices:

$$n_{m,m'}^{I,\sigma} = \sum_{\mathbf{k},\mathbf{i}} f_{\mathbf{k},\mathbf{i}}^\sigma \langle \varphi_m^{I,\sigma} | \psi_{\mathbf{k},\mathbf{i}}^\sigma \rangle \langle \psi_{\mathbf{k},\mathbf{i}}^\sigma | \varphi_{m'}^{I,\sigma} \rangle,$$

where $\varphi_m^{I,\sigma}(\mathbf{r})$ are localized (d or f) atomic orbitals

Note on onsite correction:

$$\frac{U}{2} \sum_{I,\lambda,\sigma} n_{\lambda}^{I,\sigma} (1 - n_{\lambda}^{I,\sigma})$$



→ **penalty for partial occupation of the localized orbitals**

Helps for Mott insulators, where LDA does not give a proper description.

Ex.: CoO and FeO LDA fails to predict a gap;
NiO, MnO: gap too small

For LDA + U calculations:
with pwscf:

Two modes:

- LDA + U scf calculations (U input)
 - use variable `Hubbard_U(ityp)`
`starting_ns_eigenvalue(m,ispin,I)`
- Calculate U (by LDA linear-response type of calculations with onsite perturbation U_α)
 - use variable `Hubbard_alpha(ityp)`

See Cococcioni and de Gironcoli,
cond-mat/0405160 and Phys. Rev. B 71, 035105 (2005)

References

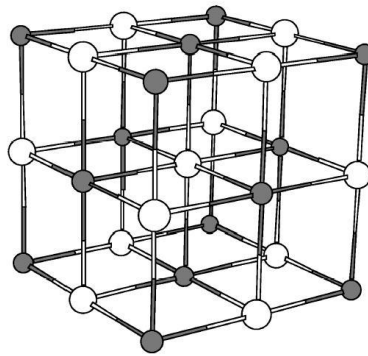
(also in INPUT_PW)

- Anisimov, Zaanen, and Andersen, PRB 44, 943 (1991)
- Anisimov et al., PRB 48, 16929 (1993)
- Liechtenstein, Anisimov, and Zaanen, PRB 52, R5467 (1994);
- Cococcioni and de Gironcoli, PRB 71, 035105 (2005)



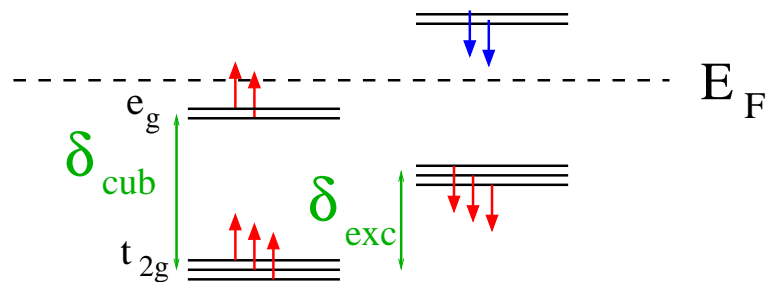
Example: NiO

- Antiferromagnetic insulator, $m_s \approx 1.7\mu_B$
- Rocksalt structure, $a = 4.17\text{\AA}$

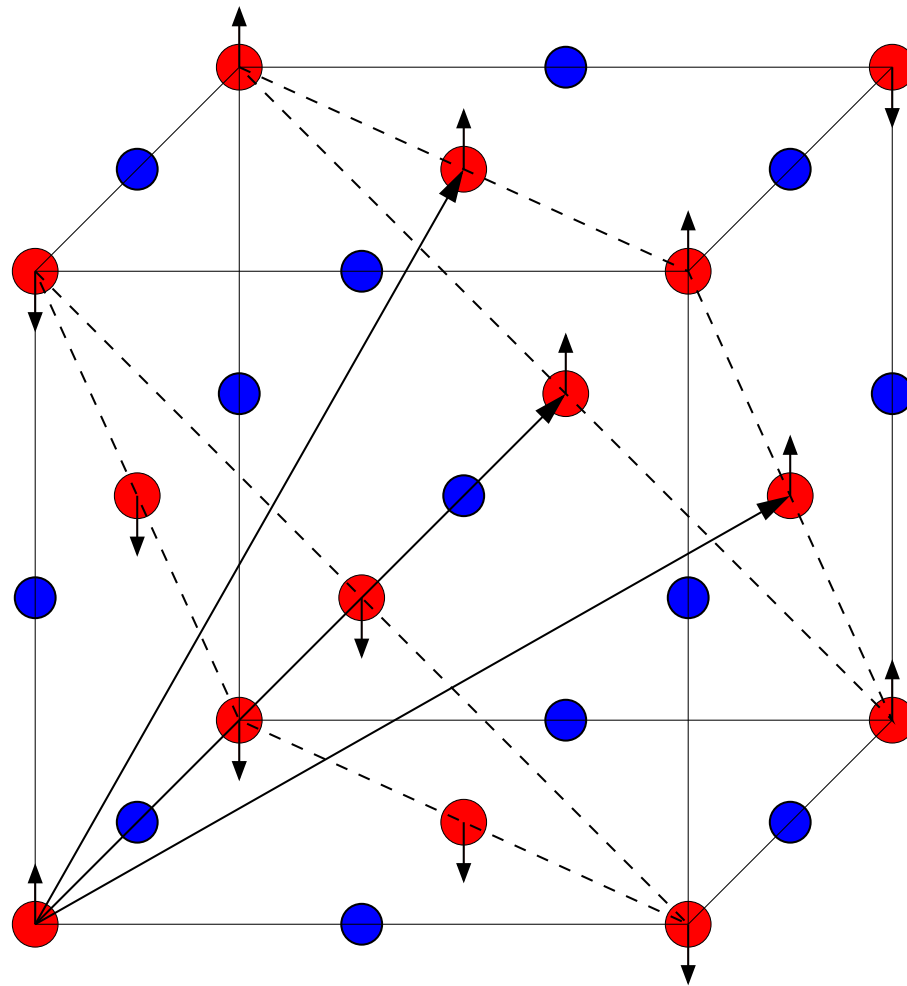


- Formal charge state: $\text{Ni}^{+2} \Rightarrow \text{Ni}(d^8)$,
with cubic crystal field:

Ni(d^8)

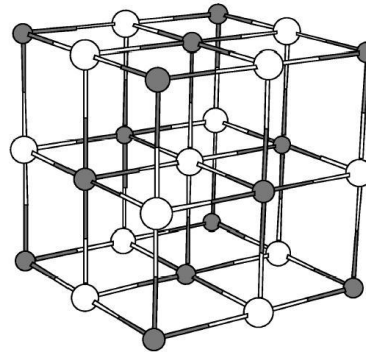


Spin structure:



Example: FeO

- Antiferromagnetic insulator, $m_s \approx 1.7\mu_B$
- Rocksalt structure, $a = 4.43\text{\AA}$



- Formal charge state: $\text{Fe}^{+2} \Rightarrow \text{Fe}(d^6)$,
with cubic crystal field:

